

Analysis of two types of polypropylene with different tacticity using msFineAnalysis Al

Product used: Mass Spectrometer (MS)

Introduction

Polypropylene has high strength, heat resistance, and excellent workability, so it is used in many industrial products. Polypropylene has a tacticity and classified into isotactic, syndiotactic and atactic depending on the arrangement of methyl group in side chains. Pyrolysis-GC-MS, which is commonly used as a polymer analysis method, detects multiple stereoisomers reflecting tacticity. In this MSTips, two types of polypropylene, isotactic and atactic, were used as samples and analyzed by the unknown compounds structure analysis software msFineAnalysis AI.



Figure 1 Tacticity of polypropylene

Experiment

Polypropylene pellets (A) isotactic and (B) atactic (=amorphous) were used as samples, and measurement was performed by the pyrolysis-GC-MS method. The sample weight was 0.2 mg. The repeat number of measurement were n = 5 times in the EI method and n = 1 time in the FI method. Qualitative information on thermal decomposition products was derived by msFineAnalysis AI, and stereoisomers were classified by the difference analysis function. Table 1 shows the measurement and analysis conditions.

Table 1 Measurement and analysis conditions

Pyrolysis conditions		MS conditions					
Pyrolyzer	EGA/PY-3030D (Frontier Lab)	Spectrometer	JMS-T2000GC (JEOL Ltd.)				
Pyrolysis Temperature	600°C	Ion Source	EI/FI combination ion source				
GC conditions		Ionization	El+:70eV, 300μA				
Gas Chromatograph	8890A GC (Agilent Technologies)		FI+:-10kV, 40mA/30msec				
Column	HP-5MS UI (Agilent Technologies)	Mass Range	<i>m/z</i> 35-800				
	30m x 0.25mm, 0.25µm	Data processing condition					
Oven Temperature	40°C(2min)-10°C/min	Software	msFineAnalysis AI (JEOL Ltd.)				
	-320ºC(60min)	Analysis mode	Variance component analysis				
Injection Mode	Split mode (50:1)		n=5 repeated measurements				
Carrier flow	He:1.5mL/min	Library database	NIST20				



TIC chromatograms of EI method

Figure 2 shows the TIC chromatogram of the EI measurement results. Chromatograms were roughly similar in (A) and (B). (A) had stronger peak intensity overall, especially for high boiling point components.



Figure 2 TIC chromatograms of EI method

Volcano plot of variance component analysis

Figure 3 shows the volcano plot of variance component analysis. Each plot corresponds to a peak on the chromatogram, and visually expresses the difference with the intensity ratio on the horizontal axis and the statistical significance (repeatability) on the vertical axis. In this analysis, 75 peaks with an intensity ratio of up to 2% to the maximum peak were targeted. All peaks were detected in both samples, but intensities differed between samples.





Results of C₁₅H₃₀ stereoisomer

The expanded TIC chromatogram is shown in Figure 4 for the three peaks (ID022-024) of the C₁₅H₃₀ stereoisomer.(A) and (B) Two chromatograms are overlaid, with peak colors based on difference analysis results. According to reference¹, the retention order of these stereoisomers was isotactic => atactic => syndiotactic. This agrees with the experimental results of this time.





Figure 5 shows the mass spectra of each peak. Since they are stereoisomers, no significant difference was observed in the mass spectra. Although the EI spectrum is not registered in the NIST library, the molecular formula of these peaks was derived to be C₁₅H₃₀ from the molecular ions detected in the FI spectrum.



Figure 5 Mass spectra of EI / FI method

Figure 6 shows the AI structure analysis results for peak ID022. Structural formula originating from polypropylene were derived with high score.



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Identification of stereoisomers by retention index and user library

Table 2 shows the qualitative analysis results by AI structural analysis only. Since AI structural analysis cannot distinguish stereoisomers, the same compound name (structural formula) was displayed for all three peaks.

General								iriance Co	mponent A	nalysis Resi	esu Total Result					
	ID	RT [min]	RI [iu]	Height [%]	Link	IM m/z	IM Ionization	Class	Log2(B/A)	p-value	Compound Name	CAS# / PubChem CID	Lib.	Similarity / Al Score	∆RI [iu]	Formula
	022	11.36	1309	30.66	√	210.23484	SI	A > B	-0.63		4,6,8,10-tetramethylundec-1-ene	19913313	Al	864	-	C15 H30
	023	11.47	1317	11.94	~	210.23424	SI	A < B	0.60	0.000	4,6,8,10-tetramethylundec-1-ene	19913313	AI	856	-	C15 H30
	024	11.59	1326	22.81	\checkmark	210.23475	SI	A = B	-0.50	0.003	4,6,8,10-tetramethylundec-1-ene	19913313	AI	860	-	C15 H30

Table 2 Results of qualitative analysis (AI structural analysis only)

The user library of msFineAnalysis AI is effective for identifying these stereoisomers. Figure 7 shows a screenshot of msFineAnalysis AI user library registration. In the user library, measured mass spectra, compound names and structural formulas derived from AI structural analysis, and retention index values can be registered. Adding the stereoisomer name to the compound name, it becomes possible to identify it in subsequent analyses.

	Add to NIST Library	×
	Compound Information	
Compound name ———	Name [Required] .6,8,10-tetramethylundec-1-ene (Isotactic) Comment	
	CAS#	
* Add stereoisomer name	Formula (Required) C15H30	
when registering		
	MW 210	
	Structure	
Potention index value	✓ with Retention Index	
	RI [iu] 1309 C=CCC(C)CC(C)CC(C)CC Draw	
	Column Type O molfille	
	Standard Non-Polar (e.g. D8-1) Semi-Standard Non-Polar (e.g. D8-5)	
	Standard Polar (e.g. DB-WAX) From NIST Library	• Structure formula
	O Not register	Structure formula
	Mass Spectrum	
	200-	
		El mass spectrum (measure
	······································	
	25 50 75 100 125 150 175 200 m/z	225
	Emar	Event Ontion
	capui	and a branch
	Target	
	NIST Library [Required] User libraly	
	Add	Cancel

Figure 7 Screenshot of user library registration

Table 3 shows the qualitative analysis results when using the user library. The retention index has made it possible to identify stereoisomers. The user library created here conforms to the NIST format and can be used with other compatible qualitative analysis software. This allows similar analyzes to be performed with low-resolution MS such as QMS.

Table 3 Results of qualitative analysis (With user	library)
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General							iriance Co	omponent A	nalysis Resu		Total Result				
ID	RT [min]	RI [iu]	Height [%]	Link	IM m/z	IM Ionization	Class	Log2(B/A)	p-value	Compound Name	CAS# / PubChem CID	Lib.	Similarity / Al Score	∆RI [iu]	Formula
022	11.36	1309	30.66	✓	210.23484	SI	A > B	-0.63	0.001	4,6,8,10-tetramethylundec-1-ene (Isotactic)	-	User libraly	999	0	C15 H30
023	11.47	1317	11.94	~	210.23424	SI	A < B	0.60	0.000	4,6,8,10-tetramethylundec-1-ene (heterotactic)	-	User libraly	999	0	C15 H30
024	11.59	1326	22.81	\checkmark	210.23475	SI	A = B	-0.50	0.003	4,6,8,10-tetramethylundec-1-ene (syndiotactic)	-	User libraly	999	0	C15 H30

Conclusion

By using JMS-T2000GC and msFineAnalysis AI, the structural formula of a polypropylene pyrolyzate could be derived, which was not registered in the NIST library. In addition, stereoisomers could be identified by using the difference analysis function and the user library. Accumulated data can be expanded to low-resolution MS such as QMS, and the JMS-T2000GC and msFineAnalysis AI set can be expected to improve the analytical capabilities of the laboratory.

Reference

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1) Shin Tsuge, Hajime Ohtani, Chuichi Watanabe (2011), Pyrolysis - GC/MS Data Book of Synthetic Polymers, Elsevier

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